

Easy access to various substituted 4-aminocyclopentenes by rearrangement of 2-ethenyl-substituted cyclopropylamines

1 PERKIN

Craig M. Williams and Armin de Meijere*

Institut für Organische Chemie, Georg-August-Universität Göttingen, Tammannstrasse 2, D-37077, Göttingen, Germany. E-mail: ameijer1@uni-goettingen.de

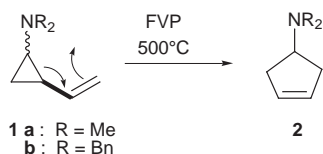
Received (in Cambridge) 8th September 1998, Accepted 5th October 1998

A variety of 2-ethenyl-substituted cyclopropylamines upon flash vacuum pyrolysis or under silver nitrate catalysis cleanly undergo ring enlargement and afford high yields (up to 95%) of 4-aminocyclopent-1-enes, some of which have unprecedented substitution patterns.

Introduction

A few biologically active compounds contain the 4-aminocyclopent-1-ene fragment,¹ yet the latter have gained real importance as precursors to nucleoside analogues with carbocyclic desoxyribose moieties.² The known synthetic routes to 4-aminocyclopent-1-enes are somewhat limited in regard to functionalisation³ and the little progress that has been made within the last decades has been centered around the parent 4-aminocyclopent-1-ene.^{1e,2b}

The well known vinylcyclopropane to cyclopentene rearrangement is one of the established routes to cyclopentenes (Scheme 1).⁴ Therefore our recently developed synthesis of



Scheme 1

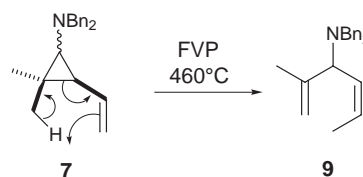
(2-ethenylcyclopropyl)amines from *N,N*-dialkylcarboxamides *via* ligand exchanged titanium-alkadiene intermediates⁵ should provide a new access to 4-aminocyclopent-1-enes especially since a dimethylamino group is known to facilitate the vinylcyclopropane rearrangement dramatically.⁶ In fact, upon attempted purification of some new *N,N*-dialkylamino-2-ethenylcyclopropanes⁵ partial rearrangement to the corresponding *N,N*-dialkylaminocyclopentenes was observed. As the previous study by Richey *et al.*⁶ was purely kinetic, and no yields were quoted, we looked into the possibility to make this rearrangement synthetically viable and applicable for substituted 4-(*N,N*-dialkylamino)cyclopent-1-enes as well.

Results and discussion

Since the dimethylamino group cannot be deprotected to a primary amino group, but benzyl groups can be removed under a variety of conditions, the 1-(*N,N*-dibenzylamino)-2-ethenyl-2-methylcyclopropane **3** (Table 1, entry 3) was investigated to insure the same rearrangement occurs with both different substituents on the amino group and the three membered ring. Initially vacuum distillation (160–170 °C/10⁻² Torr) gave only partial conversion to the expected aminocyclopentene **4** although at slightly higher temperatures and a higher pressure (170–180 °C/2–3 Torr) total conversion was achieved, and compound **4** was obtained in 60–70% yield. However, when **3** was subjected to flash vacuum pyrolysis (FVP), the product **4** was

isolated in 90% yield. In view of this success the unsubstituted 1-(*N,N*-dialkylamino)-2-ethenylcyclopropanes **1a,b** (entries 1,2) were also subjected to flash vacuum pyrolysis conditions. Initially this was somewhat problematic with the dimethylamino derivative **1a** due to the higher volatility of the substrate, however when the pressure was increased (10⁻² Torr) and the rate of evaporation of the substrate **1a** dramatically decreased, the aminocyclopentene **2a** was obtained in 90% yield. The yield is essentially quantitative, yet due to the volatility of the product **2a**, losses occurred upon small scale operation. The *N,N*-dibenzylamino derivative **1b** under the same conditions gave **2b** in 95% isolated yield (entry 2).

The rearrangement of the 1-(*N,N*-dibenzylamino)-2-ethenyl-2-(4'-methylpent-3'-enyl)cyclopropane **5** could be carried out at 400 °C, but was less efficient and gave the 1-substituted 4-dibenzylaminocyclopentene **6** only in 54% yield. With additional substituents around the three-membered ring, the rearrangement apparently tends to lead to more side products or even take a different route. For example, 1-(*N,N*-dibenzylamino)-3,3-dimethyl-2-ethenylcyclopropane **7** (entry 5), rearranged only at slightly higher temperatures (460 °C)⁷ and led to two products along with starting material. Only the minor product (6%) was the expected aminocyclopentene **8** while the major product (36%) was *cis*-3-(*N,N*-dibenzylamino)-2-methylhexa-1,4-diene **9** derived from **7** by a homo-1,5-hydrogen shift⁸ (Scheme 2). Difficulties were also encountered

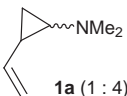
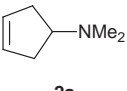
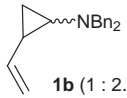
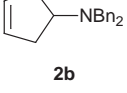
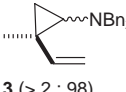
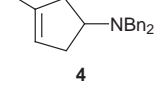
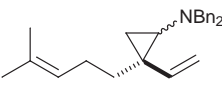
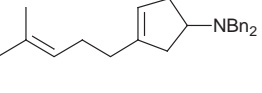
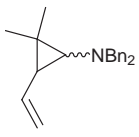
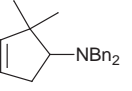
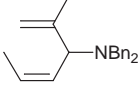
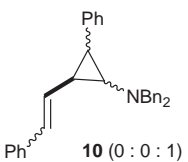
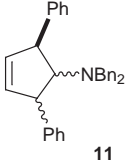
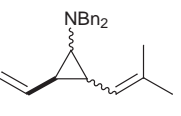
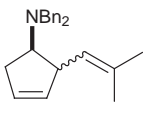
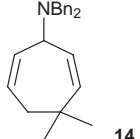
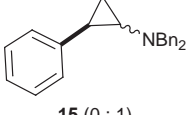
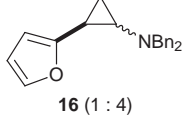
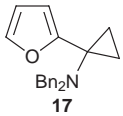


Scheme 2

with 1-(*N,N*-dibenzylamino)-2-phenyl-3-(2-phenylethenyl)cyclopropane **10** (entry 6). Due to the higher molecular mass relatively high sublimation temperatures (250 °C) were needed to introduce the material into the hot tube. This apparently led to side reactions and resulted in lower yields of product (27%). Surprisingly, the same rearrangement was induced when **10** was passed through a plug of silver nitrate-impregnated silica, to give the same aminocyclopentene **11** in slightly better yield (30%).

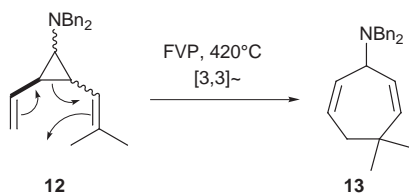
In order to test whether an amino group would facilitate the vinylcyclopropane to cyclopentene rearrangement rather than a Cope rearrangement, 1-(*N,N*-dibenzylamino)-2-ethenyl-3-(2-methylpropenyl)cyclopropane **12** (entry 7) was pyrolysed at

Table 1 Flash vacuum pyrolysis of various 2-ethenylcyclopropylamines to yield 4-aminocyclopent-1-enes

| Entry | Starting Material (<i>cis:trans</i>) | Product | Yield (%) | <i>T</i> /°C | <i>P</i> /Torr |
|-------|--|---|----------------------|--------------|-------------------|
| 1 |  1a (1 : 4) |  2a | 90 | 500 | 10 ⁻² |
| 2 |  1b (1 : 2.7) |  2b | 95 | 500 | >10 ⁻⁵ |
| 3 |  3 (> 2 : 98) |  4 | 90 | 400 | >10 ⁻⁵ |
| 4 |  5 (> 2 : 98) |  6 | 54 | 400 | >10 ⁻⁵ |
| 5 |  7 (1 : 5.3) |  8 +  9 | 6 + 36 | 460 | >10 ⁻⁵ |
| 6 |  10 (0 : 0 : 1) |  11 | 27 (30) ^a | 500 | >10 ⁻⁵ |
| 7 |  12 (1 : 2 : 2) |  13 (1 : 4) +  14 | 5 + 61 | 420 | >10 ⁻⁵ |
| 8 |  15 (0 : 1) | - ^b | 0 | 400–650 | >10 ⁻⁵ |
| 9 |  16 (1 : 4) | - ^b | 0 | 400–650 | >10 ⁻⁵ |
| 10 |  17 | - ^b | 0 | 400–650 | >10 ⁻⁵ |

^a Yield in parentheses obtained at ambient temperature under silver nitrate catalysis. ^b Starting material was recovered at 400 °C, however, increased decomposition was observed with increased temperature.

420 °C. The expected aminocyclopentene **13** was obtained in only 5% yield, the major product was 3-(*N,N*-dibenzylamino)-6,6-dimethylcyclohepta-1,4-diene **14** (61%). The structure of **14** was established on the basis of 2D NOESY and long range HMBC NMR spectra. Apparently, the [3,3]-sigmatropic (Cope) rearrangement of **12**, at least of its *cis*-2,3-dialkenyl diastereomers is faster than the vinylcyclopropane to cyclopentene rearrangement in spite of the two terminal methyl substituents in **12** (Scheme 3), and *trans*-*cis* isomerisation especially



Scheme 3

of donor-substituted ethenylcyclopropanes is known to be faster than their ring enlargement to cyclopentenenes.^{6,9}

The (arylcyclopropyl)amines 1-(*N,N*-dibenzylamino)-2-phenylcyclopropane **15**, 1-(*N,N*-dibenzylamino)-2-(2-furyl)cyclopropane **16**, and 1-(*N,N*-dibenzylamino)-1-(2-furyl)cyclopropane **17** (entries 8–10) did not undergo the expected ring-enlarging rearrangement in a temperature range of 400–650 °C; they were either recovered intact at lower or fully decomposed at higher temperatures. If 1,3-diradicals are initially formed by homolytic ring opening of cyclopropyl groups in compounds of types **15**–**17**, they apparently have little tendency to undergo five-membered ring closure by radicophilic attack on the aromatic rings.

In conclusion, several 4-aminocyclopent-1-enes are now accessible in good to high yields within two steps from conjugated dienes *via* 2-alkenyl-cyclopropylamines^{5a} and their thermal rearrangement.

Experimental

¹H (250 MHz) and ¹³C (62.9 MHz) NMR spectra were obtained with a Bruker AM 250 nuclear magnetic resonance spectrometer in deuteriochloroform with tetramethylsilane as an internal standard, coupling constants are reported in Hz. Mass spectra were recorded on a Varian MAT CH7, MAT 731 and high resolution mass spectra were recorded on a Varian MAT 311 A spectrometer with preselected molecular peak matching at $R \gg 10000$ to be within ± 2 ppm of the exact masses. Microanalyses were performed by the Mikroanalytisches Laboratorium des Instituts für Organische Chemie der Georg-August-Universität Göttingen. Melting points were determined on a Büchi 510 and are uncorrected. Flash vacuum pyrolysis experiments were conducted using a hollow ceramic electric tube furnace containing a quartz tube (71.5 cm \times 2.5 cm) placed horizontally in the centre with a flask connected at one end for sublimation of the starting material. The other end was connected to a vacuum system where the products were collected either in the exit tube or the liquid nitrogen cold trap. Pyrolysis conditions are quoted in parenthesis (sublimation oven temperature/thermolysis oven temperature/pressure). Radial chromatography was performed on a Chromatotron, Model 7924T using silica gel PF-254 with CaSO₄·0.5 H₂O type 60 for TLC, Merck 7749.

4-(*N,N*-Dimethylamino)cyclopent-1-ene **2a**

1-(*N,N*-Dimethylamino)-2-ethenylcyclopropane (491 mg, 4.42 mmol) was subjected to flash vacuum pyrolysis (ambient temp./500 °C/10⁻² Torr) with slow evaporation. The condensed product was a colourless oil (450 mg, 90%), and it was shown by ¹H NMR spectroscopy to be 99% pure, however no further

purification was attempted due to the high volatility of the product (Found: M^+ 111.1047. Calc. for C₇H₁₃N, M 111.1047). δ_H 2.13–2.33 [2 H, m, 3(5)-H], 2.21 (6 H, s, NCH₃), 2.33–2.48 [2 H, m, 3(5)-H], 2.91 (1 H, quintet, J 7.7, 4-H), 5.62 [2 H, s, 1(2)-H]; δ_C 36.48 (2 \times t), 43.41 (2 \times q), 66.40 (d), 129.13 (2 \times d); m/z 111 (M^+ , 100%), 96 (38), 85 (54), 82 (47), 70 (42), 67 (29), 58 (21), 55 (15), 53 (9).

4-(*N,N*-Dibenzylamino)cyclopent-1-ene **2b**

1-(*N,N*-Dibenzylamino)-2-ethenylcyclopropane (410 mg, 1.56 mmol) was subjected to flash vacuum pyrolysis (175 °C/500 °C/10⁻⁵ Torr). The product, a pale yellow oil, was placed on a silica plug, and this was washed with copious amounts of *n*-hexane, and the product itself was eluted with *tert*-butyl methyl ether-*n*-hexane (7:93). After evaporation of the solvents, the residue was distilled in a Kugelrohr (150–170 °C/0.01 Torr) to give **2b** as a colourless oil (390 mg, 95%) (Found: M^+ 263.1673. Calc. for C₁₉H₂₁N, M 263.1673). δ_H 2.45 [4 H, d, J 6.8, 3(5)-H], 3.57 [4 H, s, N(CH₂Ph)₂], 3.69 (1 H, quintet, J 6.8, 4-H), 5.73 [2 H, s, 1(2)-H], 7.22–7.44 (10 H, m, ArH); δ_C 34.50 (2 \times t), 54.16 (2 \times t), 57.97 (d), 126.62 (2 \times d), 128.09 (4 \times d), 128.63 (4 \times d), 129.57 (2 \times d), 140.29 (2 \times s); m/z 263 (M^+ , 20%), 234 (3), 222 (4), 186 (2), 172 (39), 156 (4), 145 (4), 130 (1), 118 (9), 106 (6), 91 (100).

4-(*N,N*-Dibenzylamino)-1-methylcyclopent-1-ene **4**

1-(*N,N*-Dibenzylamino)-2-ethenyl-2-methylcyclopropane (420 mg, 1.51 mmol) was subjected to flash vacuum pyrolysis (150–200 °C/400 °C/10⁻⁵ Torr). The pyrolysate was subjected to radial chromatography (*tert*-butyl methyl ether-*n*-hexane, 2:98–10:90 gradient) on silica gel affording compound **4** as a colourless oil (379 mg, 90%) (Found: C, 86.5; H, 8.3%; M^+ 277.1830. Calc. for C₂₀H₂₃N, C, 86.6; H, 8.4%; M 277.1830). δ_H 1.71 (3 H, br s, CH₃), 2.29–2.48 [4 H, m, 3(5)-H], 3.53 [4 H, s, N(CH₂Ph)₂], 3.66 (1 H, quintet, J 6.9, 4-H), 5.28 (1 H, br s, 1(2)-H), 7.17–7.48 (10 H, m, ArH); δ_C 16.77 (q), 34.67 (t), 38.94 (t), 54.20 (2 \times t), 58.91 (d), 123.09 (d), 126.58 (2 \times d), 128.05 (4 \times d), 128.61 (4 \times d), 138.94 (s), 140.32 (2 \times s); m/z 277 (M^+ , 42%), 262 (6), 249 (5), 224 (18), 222 (11), 200 (6), 186 (43), 181 (2), 159 (3), 132 (4), 117 (4), 106 (14), 91 (100).

4-(*N,N*-Dibenzylamino)-1-(4'-methylpent-3'-en-1'-yl)cyclopent-1-ene **6**

1-(*N,N*-Dibenzylamino)-2-ethenyl-2-(4'-methylpent-3'-en-1'-yl)cyclopropane (562 mg, 1.63 mmol) was subjected to flash vacuum pyrolysis (150–200 °C/400 °C/10⁻⁵ Torr). The pyrolysate was subjected to radial chromatography (*tert*-butyl methyl ether-*n*-hexane, 2:98–10:90 gradient) on silica gel affording **6** as a colourless oil (301 mg, 54%) (Found: C, 87.2; H, 9.2%; M^+ 345.2456. Calc. for C₂₅H₃₁N, C, 86.9; H, 9.0%; M 345.2456). δ_H 1.64 (3 H, s, CH₃), 1.71 (3 H, s, CH₃), 2.02–2.27 (4 H, m, CH₂CH₂), 2.30–2.54 [4 H, m, 3(5)-H], 3.58 [4 H, s, N(CH₂Ph)₂], 3.69 (1 H, quintet, J 6.9, 4-H), 5.08–5.21 (1 H, m, 2-H), 5.35 [1 H, br s, CH=C(CH₃)₂], 7.15–7.54 (10 H, m, ArH); δ_C 17.69 (q), 25.68 (q), 26.27 (t), 31.45 (t), 34.44 (t), 37.21 (t), 54.18 (2 \times t), 58.46 (d), 122.14 (d), 124.31 (d), 126.59 (2 \times d), 128.07 (4 \times d), 128.64 (4 \times d), 131.45 (s), 140.39 (2 \times s), 143.26 (s); m/z 345 (M^+ , 35%), 313 (1), 276 (42), 254 (8), 239 (2), 224 (27), 222 (16), 210 (17), 181 (5), 132 (2), 146 (1), 117 (4), 106 (8), 91 (100).

4-(*N,N*-Dibenzylamino)-3,3-dimethylcyclopent-1-ene **8** and *cis*-3-(*N,N*-dibenzylamino)-2-methylhexa-1,4-diene **9**

1-(*N,N*-Dibenzylamino)-3,3-dimethyl-2-ethenylcyclopropane (263 mg, 0.9 mmol) was subjected to flash vacuum pyrolysis (250 °C/460 °C/10⁻⁵ Torr). The pyrolysate was subjected to column chromatography (ethyl acetate-*n*-hexane, 10:90) on silver nitrate-impregnated silica gel¹⁰ affording three fractions.

The first fraction contained *cis*-3-(*N,N*-dibenzylamino)-2-methylhexa-1,4-diene **9** as a colourless oil (94 mg, 36%) (Found: M^+ 291.1986. Calc. for $C_{21}H_{25}N$, M 291.1986). δ_H 1.43 (3 H, dd, J 6.8 and 1.6, CH_3), 1.92 (3 H, s, CH_3), 3.61 [4 H, AB, J 14.0, $N(CH_2Ph)_2$], 3.91 (1 H, d, J 9.7, 3-H), 4.83–4.98 (2 H, m, $CH_2=$), 5.61–5.73 (1 H, m, 5-H), 5.79–5.94 (1 H, m, 4-H), 7.18–7.60 (10 H, m, ArH); δ_C 13.29 (q), 21.12 (q), 53.57 (2 \times t), 60.18 (d), 112.76 (t), 126.23 (d), 126.60 (2 \times d), 128.09 (4 \times d), 128.26 (d), 128.64 (4 \times d), 140.25 (2 \times s), 146.21 (s); m/z 291 (M^+ , 6%), 250 (100), 196 (1), 182 (6), 158 (2), 118 (1), 104 (2), 95 (12), 91 (68). The second fraction contained the cyclopentene **8** as a colourless oil (15 mg, 6%) (Found: M^+ 291.1986; $C_{21}H_{25}N$ requires M 291.1986). δ_H 0.88 (3 H, s, CH_3), 1.20 (3 H, s, CH_3), 2.28–2.57 (2 H, m, 5-H), 3.09 (1 H, dd, J 7.5 and 5.1, 4-H), 3.63 [4 H, AB, J 14.4, $N(CH_2Ph)_2$], 5.51–6.66 [2 H, m, 1(2)-H], 7.18–7.47 (10 H, m, ArH); δ_C 23.37 (q), 29.39 (q), 30.02 (t), 47.40 (s), 55.64 (2 \times t), 67.54 (d), 126.22 (d), 126.57 (2 \times d), 128.11 (4 \times d), 128.42 (4 \times d), 140.62 (d), 142.02 (2 \times s); m/z 291 (M^+ , 39%), 276 (11), 250 (39), 222 (17), 200 (21), 174 (5), 145 (3), 132 (25), 106 (4), 95 (10), 91 (100). The third fraction contained starting material (47 mg, 18%).

4-(*N,N*-Dibenzylamino)-3,5-diphenylcyclopent-1-ene **11**

1-(*N,N*-Dibenzylamino)-2-phenyl-2-(2-phenylethenyl)cyclopropane (164 mg, 0.39 mmol) was passed through a column (ethyl acetate–*n*-hexane, 1:19) of silver nitrate-impregnated silica gel¹⁰ affording the title compound **11** as a colourless oil (50 mg, 30%) (Found: M^+ 415.2299. Calc. for $C_{31}H_{29}N$, M 415.2299). δ_H 3.47 (1 H, t, J 8.0, 4-H), 3.72 [4 H, s, $N(CH_2Ph)_2$], 4.13 [2 H, d, J 8.0, 3(5)-H], 5.82 [2 H, s, 1(2)-H], 6.90–6.95 (4 H, m, ArH), 7.05–7.15 (6 H, m, ArH), 7.18–7.37 (10 H, m, ArH); δ_C 52.19 (d), 55.01 (2 \times t), 78.63 (2 \times d), 126.19 (2 \times d), 126.42 (2 \times d), 127.79 (4 \times d), 128.07 (4 \times d), 128.26 (4 \times d), 128.31 (4 \times d), 134.19 (2 \times d), 139.38 (2 \times s), 144.90 (2 \times s); m/z 415 (M^+ , 100%), 338 (4), 324 (28), 310 (7), 298 (12), 279 (4), 246 (3), 218 (10), 206 (6), 193 (17), 167 (7), 149 (8), 117 (21), 91 (96).

1-(*N,N*-Dibenzylamino)-2-phenyl-2-(2-phenylethenyl)cyclopropane (104 mg, 0.25 mmol) was subjected to flash vacuum pyrolysis (250 °C/500 °C/ $>10^{-5}$ Torr). The pyrolysate was twice subjected to radial chromatography (*tert*-butyl methyl ether–*n*-hexane, 5:95) to give the title compound **11** (28 mg, 27%).

4-(*N,N*-Dibenzylamino)-3-(2-methylprop-1-enyl)cyclopent-1-ene **13** and 3-(*N,N*-dibenzylamino)-6,6-dimethylcyclohepta-1,4-diene **14**

1-(*N,N*-Dibenzylamino)-2-ethenyl-3-(2-methylprop-1-enyl)-cyclopropane **12** [two major diastereomers only, however, a (1:2:2) mixture of all three gave the same result] (82 mg, 0.26 mmol) was subjected to flash vacuum pyrolysis (250 °C/420 °C/ $>10^{-5}$ Torr). The pyrolysate was subjected to column chromatography (ethyl acetate–*n*-hexane, 10:90) on silver nitrate-impregnated silica gel¹⁰ affording two fractions. The first fraction contained 3-(*N,N*-dibenzylamino)-6,6-dimethylcyclohepta-1,4-diene **14** as a white solid, which was recrystallised from *n*-hexane (–23 °C), mp 72–73 °C (50 mg, 61%) (Found: M^+ 317.2143. Calc. for $C_{23}H_{27}N$, M 317.2143). δ_H 0.99 (3 H, s, CH_3), 1.02 (3 H, s, CH_3), 1.95 (1 H, dd, J 14.2 and 7.0, 7-H), 2.28 (1 H, dd, J 14.2 and 6.0, 7-H), 3.67 [4 H, AB, J 14.1, $N(CH_2Ph)_2$], 4.17 (1 H, br s, 3-H), 5.43–5.58 [2 H, m, 2(4)-H], 5.79–5.89 (1 H, m, 1-H), 5.99–6.05 (1 H, m, 5-H), 7.20–7.43 (10 H, m, ArH); δ_C 28.86 (q), 31.10 (q), 34.99 (s), 39.61 (t), 54.12 (2 \times t), 56.53 (d), 126.71 (2 \times d), 127.13 (d), 128.15 (4 \times d), 128.52 (4 \times d), 129.16 (d), 133.07 (d), 140.25 (d), 141.52 (2 \times s);

m/z 317 (M^+ , 4%), 302 (6), 226 (18), 197 (11), 181 (1), 170 (3), 148 (1), 132 (1), 120 (8), 106 (11), 91 (100). The second fraction contained compound **13** (mixture of two diastereomers, ratio 4:1) as a colourless oil (4 mg, 5%) (Found: M^+ 317.2143. Calc. for $C_{23}H_{27}N$, M 317.2143). (Only the major diastereomer) δ_H 1.71 (3 H, s, CH_3), 1.77 (3 H, s, CH_3), 2.33–2.52 (2 H, m, 5-H), 3.28–3.37 (1 H, m, 4-H), 3.50–3.63 (1 H, s, 3-H), 3.59 [4 H, s, $N(CH_2Ph)_2$], 4.80–4.86 [1 H, m, $CHC(CH_3)_2$], 5.42–5.50 (1 H, m, 1-H), 5.62–5.79 (1 H, m, 2-H), 7.18–7.40 (10 H, m, ArH); δ_C 18.30 (q), 25.75 (q), 33.95 (t), 45.59 (d), 54.49 (2 \times t), 65.72 (d), 126.59 (2 \times d), 128.03 (4 \times d), 128.54 (d), 128.56 (4 \times d), 129.07 (d), 130.42 (s), 133.80 (d), 140.41 (2 \times s); m/z 317 (M^+ , 49%), 302 (8), 288 (3), 248 (4), 246 (9), 226 (11), 222 (12), 206 (2), 196 (1), 158 (6), 132 (7), 106 (6), 91 (100).

Acknowledgements

This work was supported by the Deutsche Forschungsgemeinschaft (SFB 416, Project A3), the Fonds der Chemischen Industrie, and by the companies BASF AG, Bayer AG, Chemetall GmbH, Degussa, and Hüls AG (chemicals). C. M. W. is indebted to the Alexander von Humboldt Foundation for a postdoctoral Research Fellowship. The authors thank Dr B. Knieriem for his careful reading of the manuscript.

References

- (a) D. Bouzard, P. Di Cesare, M. Essiz, J. P. Jacquet, P. Remuzon, A. Weber, T. Oki and M. Masuyoshi, *J. Med. Chem.*, 1989, **32**, 537; (b) A. Piergentili, P. Angeli, M. Giannella, M. Pignini, W. Quaglia and S. K. Tayebati, *Arzneim. Forsch.*, 1996, **46**, 99; (c) D. F. Morrow, P. C. Johnson, H. Torabi, D. Williams, D. L. Wedding, J. W. Craig and R. F. Majewski, *J. Med. Chem.*, 1973, **16**, 736; (d) S. Kamata, N. Haga, T. Tsuru, K. Uchida, H. Kakushi, H. Arita and K. Hanasaki, *J. Med. Chem.*, 1990, **33**, 299; (e) D. P. Curran, S. A. Gothe and S.-M. Choi, *Heterocycles*, 1993, **35**, 1371.
- (a) K. C. Murdock and R. B. Angier, *J. Org. Chem.*, 1962, **27**, 3317; (b) R. D. Elliot, G. A. Rener, J. M. Riordan, J. A. Secrist III, L. L. Bennett, Jr., W. B. Parker and J. A. Montgomery, *J. Med. Chem.*, 1994, **37**, 739; (c) M. Koga and S. W. Schneller, *Tetrahedron Lett.*, 1990, **31**, 5861; (d) K. C. Murdock and R. B. Angier, *Tetrahedron Lett.*, 1962, 415; (e) J. Béres, Gy. Sági, I. Tömösközi, L. Gruber, E. Baitz-Gács, L. Ötvös and E. De Clercq, *J. Med. Chem.*, 1990, **33**, 1353; (f) M. Legraverend, C. Huel, J. Guilhem and E. Bisagni, *Carbohydr. Res.*, 1992, **228**, 21; (g) F. Girard, C. Demaison, M.-G. Lee and L. A. Agrofoglio, *Tetrahedron*, 1998, **54**, 8745.
- (a) K. C. Murdock and R. B. Angier, *J. Org. Chem.*, 1962, **27**, 2395; (b) K. Hammer and K. Undheim, *Tetrahedron*, 1997, **53**, 2309.
- T. Hudlicky, D. A. Becker, R. L. Fan and S. Kozhushkov, in *Methods of Organic Chemistry* (Houben-Weyl), ed. A. de Meijere, Thieme, Stuttgart, 1997, vol. E 17c, p. 2538.
- (a) C. M. Williams, V. Chaplinski, P. R. Schreiner and A. de Meijere, *Tetrahedron Lett.*, 1998, **39**, 7695; (b) A. de Meijere, A. Kourdioukov, V. Chaplinski, M. Kordes and C. M. Williams, unpublished work.
- H. G. Richey, Jr. and D. W. Shull, *Tetrahedron Lett.*, 1976, 575.
- Even higher temperatures resulted in decomposition and lower temperatures afforded more starting material.
- H. M. Frey and R. K. Solly, *J. Chem. Soc. (B)*, 1970, 996.
- (a) J. M. Simpson and H. G. Richey, *Tetrahedron Lett.*, 1973, 2545; (b) G. McGaffin, B. Grimm, U. Heinicke, H. Michaelsen, A. de Meijere and R. Walsh, unpublished work.
- L. J. Morris, *Chem. Ind.*, 1962, 1239; silica gel (200 g) was suspended in acetonitrile (~400 ml) to which was added a solution of silver nitrate (40 g) in acetonitrile (~50 ml). The solvent was then slowly evaporated (1 Torr) and the silica gel placed in an oven (80 °C) or in an evacuated flask (0.01 Torr) for 24 h.